$\delta^{13}C$ and $\delta^{15}N$ isotopic analysis of solid materials

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Overview:

Analyses of δ^{13} C, wt%C, δ^{15} N and wt%N of solid matter are done using Continuous Flow-Elemental Analysis-Isotope Ratio Mass Spectrometry (CF-EA-IRMS) technology. In the ISL-AGg, a Thermo DeltaV^{Plus®} mass spectrometer is interfaced with either an Elementar *Isotope CUBE*® or Costech 4010[®] elemental analyzer via a ConfloIV[®] device. All materials (RMs, QA/QCs and unknowns) are packed in tin cups of varying and appropriate size, which are dropped by auto sampler onto a guartz tube combustion reactor. The temperature of this reactor is maintained at ~1000°C and 'flash-combustion' is achieved by injecting a pulse of $O_2(gas)$ exactly at the time of sample drop. The eluent gases are then swept by the helium carrier stream through a reduction reactor (~650°C), thus reducing NO_x species to $N_{2(gas)}$. GC separation of N₂ and CO₂ is achieved before the gas stream is leaked through the Conflo-IV open split into the ion source of the mass analyzer. δ^{13} C and δ^{15} N values are determined by comparing the respective sample peak areas, as [Vs], to reference gas peaks also inlet through the open split. For materials with widely varying [C:N] ratios, δ^{13} C and δ^{15} N analyses must be done separately. For samples whose C:N ratio approaches [3:1] (i.e. bone collagen) the peak jumping feature of the mass spectrometer can be used and a single sample suffices. These instruments are fully automated and computer controlled using ISODAT 3.88 software. Prior to sample analysis instrument conditions: reactor temperature(s), carrier gas flows, ion beam background(s), ion source stability and signal linearity are checked and recorded ("Daily check" routine).

Stable isotope ratios are expressed as delta (δ) and are measures of a 'per mill' (‰), or parts per thousand difference between the isotope ratio of a sample and that of a known (International) standard material. Values are reported relative to 'Vienna Peedee Belemnite' (VPDB) formation for Carbon (Craig, 1957) and 'Atmospheric air' for Nitrogen (Air-N₂).

Internal lab standards:

Standard	Supplier	Prod. #	Formula
Caffeine	Sigma Aldrich	C-0750	C8H10N4O2
Gelatin	Sigma Aldrich	G-9382	unknown
Glycine	MP Biomedicals	100570	C2H5NO2
Keratin	MP Biomedicals	902111	unknown

International standards:

<u>Identifier</u>	δ ¹³ C (‰)vpdb_	δ ¹⁵ N (‰)Air-N2
USGS 24	-16.0 ± 0.1	
IAEA-CH-6	-10.4 ± 0.2	
IAEA-CH-7	-31.8 ± 0.2	
NBS 22	-30.03 ± 0.2	
USGS 40	-26.39 ± 0.2	-4.52 ± 0.2
USGS 41	37.63 ± 0.2	+47.57 ± 0.2
USGS 25		-30.40 ± 0.2
USGS 26		+53.70 ± 0.2
USGS 34		-1.80 ± 0.2
USGS 35		+2.70 ± 0.2
IAEA N1		+0.43 ± 0.2
IAEA N2		+20.32 ± 0.2
IAEA NO3		$+4.69 \pm 0.2$

- Internal lab standards are used at the beginning, between (~ every 5th) and the end of each sequence to correct for instrument drift and to normalize the data to internationally accepted standards
- Six (6) replicates of different weights are placed near the beginning of each sequence to allow for element wt.% determination and 'non-linearity' correction
- Internal lab standards have been characterized against the International Standards listed above and are re-checked periodically
- USGS LIMs is used for drift correction, normalization and data management. (http://water.usgs.gov/software/LIMS/)

Accuracy and Precision:

 δ^{13} Corganic ± 0.2 per mil (n=10 internal lab standards) δ^{15} Norganic ± 0.2 per mil (n=10 internal lab standards) Elemental wt%C, wt%N and C/N ratio = ± 5%

References:

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