# $\delta^{15}$ N and $\delta^{18}$ O isotopic analysis by "*bacterial denitrifier*" method

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## Overview:

A method used to determine the N and O isotopic composition of 'Total Oxidized Nitrogen' (TON) in fresh or sea waters where (TON =  $NO_3^- + NO_2^-$ ). Note: most surface and groundwaters under '*natural*' conditions contain dissolved nitrite << dissolved nitrate, such that dissolved TON  $\approx$  dissolved nitrate and is used here interchangeably.

The 'bacterial denitrifier' method is based on bacterial reduction of TON to nitrous oxide gas (N<sub>2</sub>O) via a bacterium that lacks nitrous oxide reductase, so further reduction to N<sub>2</sub> does not occur. This enables simultaneous determination of both  $\delta^{15}N$  and  $\delta^{18}O$  values of the sample NO<sub>3</sub><sup>-</sup> by measuring  $\delta^{15}N$  and  $\delta^{18}O$  values of the produced N<sub>2</sub>O<sub>(g)</sub>. The bacterial strain *Pseudomonas aureofaciens* (ATCC# 13985) is used to reduce TON to N<sub>2</sub>O<sub>(g)</sub>.

#### **Dissolved TON extraction:**

Briefly, *P.aureofaciens* bacteria are grown in a special Tryptic Soy Broth (TSB) prepared in advance. Batches of TSB are inoculated using sterile techniques and allowed 7 days of growth, after which time the bacteria will have reached maximum population density and exhausted the TSB nutrients. The bacteria are then harvested and subdivided into individual 20ml, crimp sealed sample vials (pre-flushed with inert N<sub>2</sub> (UHP 4.8) for ~3-4 hours to lower the atmospheric blank). Standards and raw sample waters are then injected into individual vials in a preset order and left overnight at room temperature. Note: the target amount of NO<sub>3</sub><sup>-</sup> depends on the sensitivity of the mass spec. but is generally in the range of 20 nmol [NO<sub>3</sub><sup>-</sup>-NO<sub>3</sub><sup>-</sup>]. The next day, each vial is lysed by injecting 0.15mL of 10 N NaOH and the vials mounted in an autosampler for automated analysis.

The analytical system is comprised of an autosampler + PreCon<sup>®</sup> + HP 6890<sup>®</sup> gas chromatogram + Conflo-IV<sup>®</sup> and Thermo DeltaVPlus<sup>®</sup> mass spectrometer. The entire instrumental process is automated, and computer controlled using ISODAT 3.88 software. Prior to sample analysis, instrument conditions (oven temperatures, carrier gas flows, ion beam background(s), ion source stability and signal linearity) are checked and recorded by the operator as part of the "*Daily check*" routine.

The instrumental technique is as follows: the sample/broth solution in a vial is sparged, and the headspace of the vial flushed, with UHP 5.0 helium at a flow rate of 15 ml/min using a custom concentric needle. The N2O<sub>(g)</sub> + carrier passes through a series of traps: alcohol/slush  $\rightarrow$  Ascarite<sup>®</sup>/H<sub>2</sub>O  $\rightarrow$  Tekmar purge/trap F<sup>®</sup> (Supelco p/n: 20293)) to remove:

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excess moisture, CO<sub>2</sub> and VOCs respectively before being cryo-focused in the PreCon. The N<sub>2</sub>O<sub>(g)</sub>, once released from the PreCon, passes through a GC column (Restek Rt-Q-BOND, 30m x 320µmID x 10µM) operated isothermally at 40°C to separate N<sub>2</sub>O from any remaining CO<sub>2</sub>, then into the open split interface of the Conflo-IV which leads to the ion source of the mass spectrometer. The 'raw'  $\delta^{15}$ N and  $\delta^{18}$ O values are calculated by the ISODAT software.

Stable isotope ratios are expressed as delta ( $\delta$ ) and are measures of a 'per mill' (‰), or parts per thousand difference between the isotope ratio of a sample and that of a known (International) standard material. Values are reported relative to '*VSMOW*' for Oxygen and '*Atmospheric air*' for Nitrogen.

## Data correction:

lon currents of masses 44, 45 and 46 are measured simultaneously and the <sup>15/14</sup>Nitrogen and <sup>18/16</sup>Oxygen ratios of the sample are compared to that of a working-reference N<sub>2</sub>O gas (4.8, 'Semiconductor Process Gas', Praxair Air, Canada). Raw data is corrected to the International 'air-N<sub>2</sub>' and 'VSMOW' scales by normalization with reference materials analyzed in the same sequence as the samples (the "IT" principal described in Werner & Brand, 2001). An <sup>17</sup>O correction is applied to the  $\delta^{15}$ N calculation by the ISODAT software.

## Reference materials:

Identifier	<u>δ<sup>15</sup>Ν (</u> ‰)Air-N2	<u>δ18</u> Ο (‰)∨sмow
IAEA NO3	+4.7 ± 0.2	+25.6 ± 0.4
USGS 32	+180 ± 1.0	+25.7 ± 0.4
USGS 34	-1.8 ± 0.2	-27.9 ± 0.6
USGS 35	n/a*	+57.5 ± 0.6

\* The non-mass dependent <sup>18</sup>O:<sup>17</sup>O:<sup>16</sup>O ratios of USGS35, a commercially purified NaNO<sub>3</sub> from the natural nitrate ore deposits of the Atacama Desert, preclude its use in  $\delta^{15}$ N normalization calculations. See: Bohlke et al. (2003).

## Accuracy and Precision:

Accuracy and precision of  $\delta^{15}N$  determinations based on a long-term record of in-house references is: 0.3 per mil

Accuracy and precision of  $\delta^{18}$ O determinations based on a long-term record of in-house references is: 0.7 per mil

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#### References:

Sigman, D.M., K.L. Casciotti, M. Andreani, C. Barford, M. Galanter, and J.K. Böhlke. A bacterial method for the nitrogen isotopic analysis of nitrate in seawater and freshwater. Anal. Chem., 2001, 73: 4145-4153.

Casciotti, K.L., D.M. Sigman, M. Galanter Hastings, J.K. Böhlke, and A. Hilkert, Measurement of the oxygen isotopic composition of nitrate in seawater and freshwater using the denitrifier method. Anal. Chem., 2002, 74: 4905-4912.

Böhlke, J.K., Mroczkowski, S.J. and T.B. Coplen, Oxygen isotopes in nitrate: new reference materials for <sup>18</sup>O: <sup>17</sup>O: <sup>16</sup>O measurements and observations on nitrate-water equilibration, Rapid Communications in Mass Spectrometry, 2003, 17; 1835-1846

Rock, L., and Ellert, B., Nitrogen-15 and Oxygen-18 Natural Abundance of Potassium Chloride Extractable Soil Nitrate Using the Denitrifier Method, Soil Science Society of America Journal, 2007, vol. 71 (2), 355-361.

McIlvin, M.R. and K.L. Casciotti, Technical Updates to the Bacterial Method for Nitrate Isotopic Analyses. Anal. Chem., 2011, 83: 1850-1856

Weigand M. A. et al., Updates to instrumentation and protocols for isotopic analysis of nitrate by the denitrifier method, Rapid Commun. Mass Spectrom. 2016, 30, 1365–1383

R.A. Werner and W.A. Brand, Referencing strategies and techniques in stable isotope ratio analysis, Rapid Communications in Mass Spectrometry, 2001: 15: 501-519